

INTERACTIONS IN COMPOSITES OF METHYLENE BLUE DYE WITH MoS_2 AND C_{60} FULLERENE CHARACTERIZED BY LASER DESORPTION/IONIZATION MASS SPECTROMETRY

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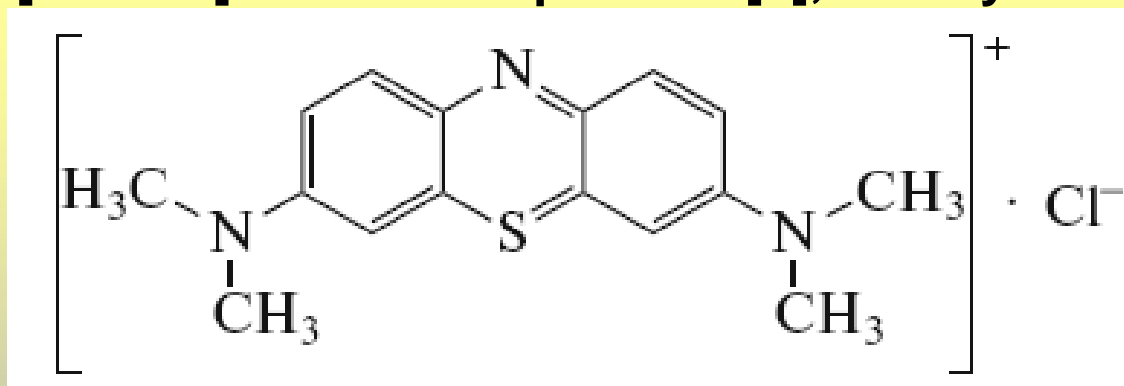
Among a variety of nanocomposites, in the present work, we studied a combination of methylene blue (MB) dye with molybdenum disulfide MoS_2 and fullerene C_{60} . The unification of several biologically active components is currently tested for the combined therapy of cancer with the expectation of a synergistic effect of different mechanisms of action: separately, MB is used for photodynamic therapy, while MoS_2 is promising for photothermal therapy. Biological effects of C_{60} are also a subject of intense research. This determines the importance of studying molecular processes, possible covalent and noncovalent interactions in composites of these substances.

Two-component ($\text{MB} + \text{MoS}_2$), ($\text{C}_{60} + \text{MoS}_2$), ($\text{MB} + \text{C}_{60}$), and three-component ($\text{MB} + \text{MoS}_2 + \text{C}_{60}$) systems were prepared by ultrasound treatment of aqueous mixtures of the components. A method of choice for studying intermolecular interactions in the multi-component systems is laser desorption/ionization (LDI) mass spectrometry.

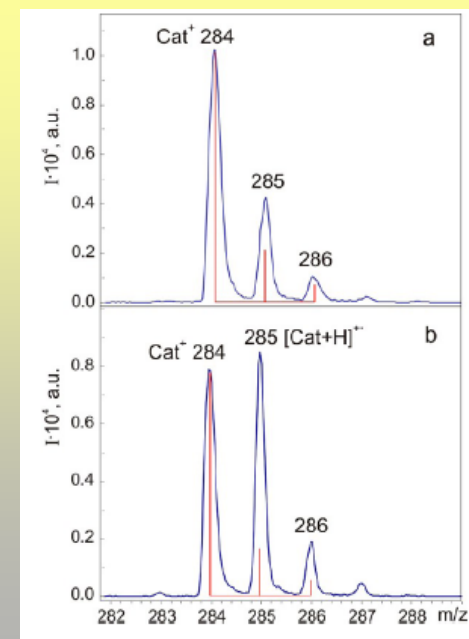


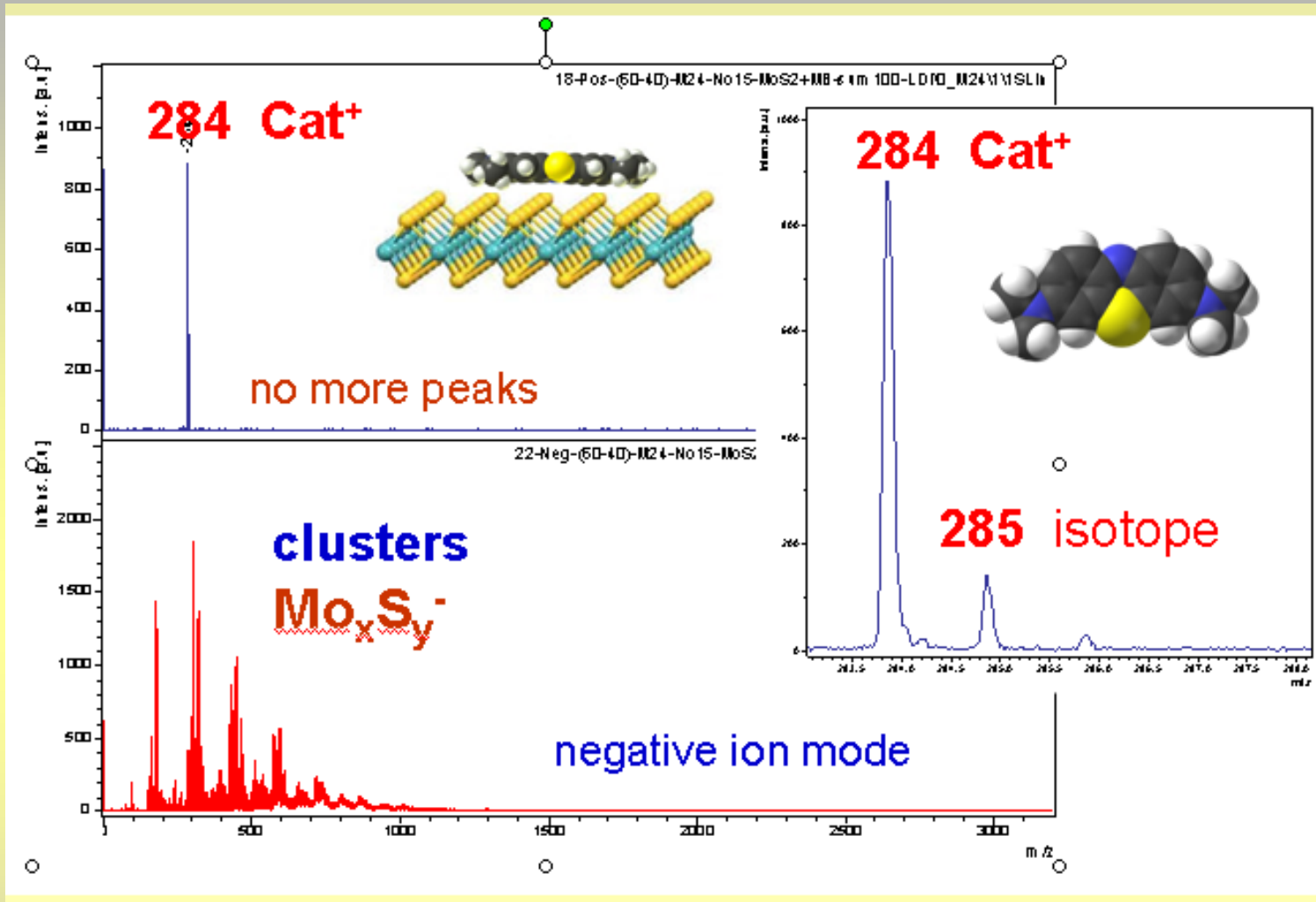
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Earlier, we have established correlations between the aggregation state and intermolecular interactions of the MB dye with characteristic features of its mass spectra [1, 2], which we were guided by in our present studies. In brief, it was established that in the case of monomeric adsorption of MB^+ cations on inert substrates, such as carbon nanotubes [1], an abundant signal of MB^+ cation is recorded in the desorption mass spectra. In the case of MB dimerization or aggregation, a redox reaction occurs, resulting in the formation of the $[\text{MB} + \text{H}]^+$ reduction product [2], readily detected in the mass spectra.



1. V.V. Chagovets, M.V. Kosevich, S.G. Stepanian, O.A. Boryak, V.S. Shelkovsky, V.S. Leontiev, V.A. Pokrovskiy, V.A. Karachevtsev, *J Phys Chem C*, **116**(38) 20579 (2012).
2. V.S. Shelkovsky, M.V. Kosevich, O.A. Boryak, V.V. Chagovets, I.V. Smigol, V.A. Pokrovskiy, *RSC Advances*, **4**(104), 60260 (2014).

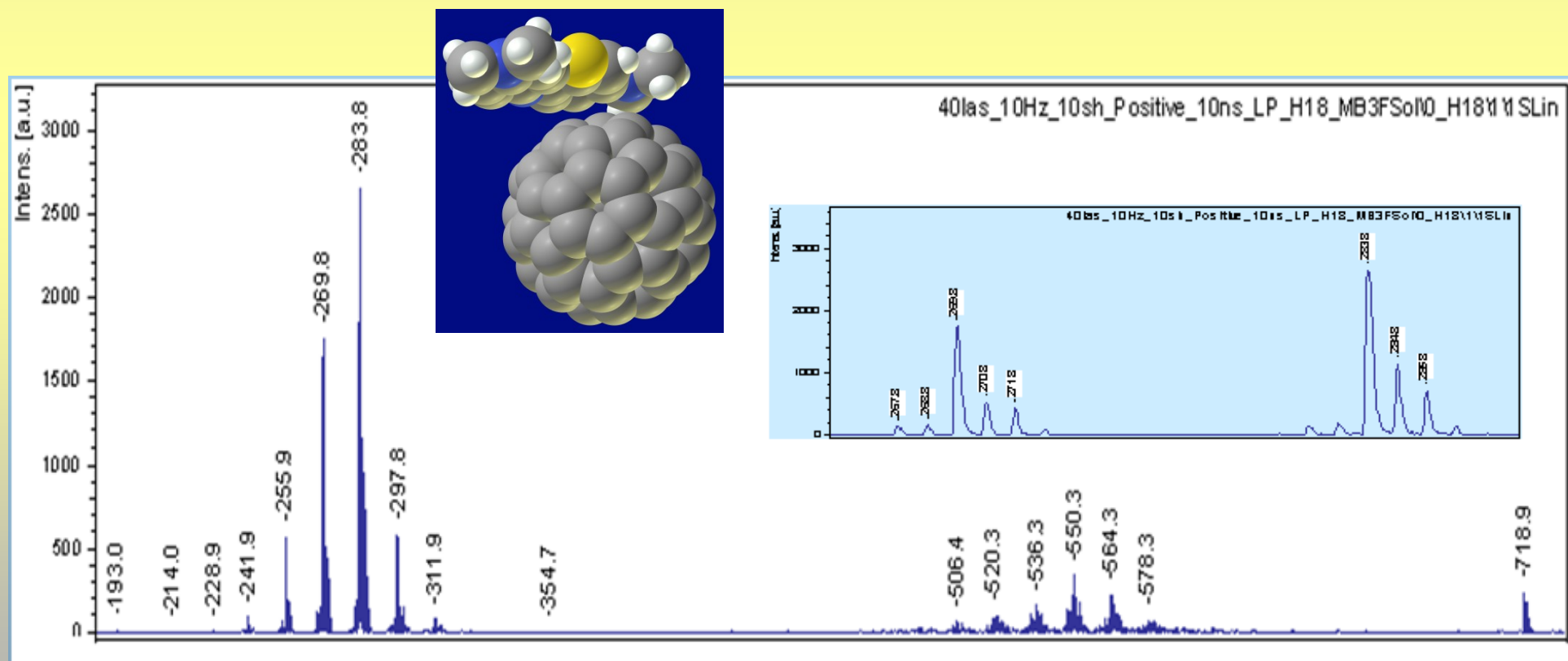




Binary (MB+MoS₂) system

In the positive ion LDI mass spectra of binary (MB + MoS₂) nanocomposite, an abundant signal of MB⁺ cation was recorded, which pointed to monomeric adsorption of the MB⁺ cation at the partially negative MoS₂ surface. Similarly to the case with carbon nanotubes, noncovalent electrostatic interactions took place.

In the LDI mass spectra of the binary (MB + C₆₀) system, substantial “damage” of MB⁺ was observed: there was a set of peak groups corresponding to sequential loss of four methyl –CH₃ groups present in the MB⁺ structure. In the higher mass range, there was a set of peak groups which may be tentatively attributed to MB dimer and species related to loss and addition of CH₃ groups. Obviously, these transformations were caused by redox interactions of MB⁺ with C₆₀^{•-}.



This effect of C_{60} was preserved in the triple (MB + MoS_2 + C_{60}) composite: a similar set of MB^+ fragments formed due to $-CH_3$ loss was registered along with the peaks characteristic of C_{60} and MoS_2 . The products of MB^+ polymerization, however, were not observed obviously because of the relatively low content of MB^+ at the MoS_2 surface.

